Oxidative Ring-Opening of *rel*-(2*R*,3*R*,5*S*)-5-Aryl-2-benzoylamino-6,7-bis(methoxycarbonyl)-2,3-dihydro-1-oxo-3-phenyl-1*H*,5*H*-pyrazolo[1,2-*a*]-pyrazoles. Synthesis of *rel*-(2*R*,3*R*)-3-Phenyl-3-[5-aryl-3,4-bis(methoxycarbonyl)pyrazolyl-1]alanine Esters

Andrej Prešeren, Jurij Svete*, and Branko Stanovnik*

Faculty of Chemistry and Chemical Technology, University of Ljubljana, Ljubljana, Slovenia Received November 23, 1998

rel-(2R,3R)-N-Benzoylamino-6,7-bis(methoxycarbonyl)-2,3-dihydro-1-oxo-1H,5H-pyrazolo[1,2-a]-pyrazoles 5, accesible by cycloaddition of dimethyl acetylenedicarboxylate (3) to (1Z)-rel-(4R,5R)-1-aryl-methylidene-4-benzoylamino-5-phenyl-3-pyrazolidinone-1-azomethine imines 4, undergo oxidative ring cleavage with methanolic bromine giving rel-(2R,3R)-N-benzoyl-3-phenyl-3-[5-aryl-3,4-bis(methoxy-carbonyl)pyrazolyl-1]alanine methyl esters 6 as products.

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Due to their occurance in nature and biological activity, several synthetic approaches for the preparation of heteroarylalanines have been developed in last few decades [1]. Among them, considerable attention has been payed to the synthesis of 3-(pyrazolyl-1)alanine (1), which was isolated from the semen of *Citrullus vulgaris* and which is so far the only naturally occuring amino acid with a pyrazolyl residue [2-7]. 3-(Pyrazolyl-1)alanine (1) was also used as constituent of highly potent renine inhibitors 2a,b [8, 9] (Figure 1).

Azomethine imines 4 and cycloadducts 5 were prepared according to the procedure described previously [10]. Treatment of 5 with methanolic bromine gave the corresponding rel-(2R,3R)-N-benzoyl-3-phenyl-3-[5-aryl-3,4-bis(methoxycarbonyl)pyrazolyl-1]alanine methyl esters 6. Presumably, the reaction mechanism could procede via bromination of pyrazolo[1,2-a]pyrazole system, followed by ring opening and elimination of hydrogen bromide. The initial bromination can occur either at the allylic

Previously, we have reported on the preparation of substituted rel-(2R,3R,5S)-5-aryl-6,7-bis(methoxycarbonyl)-2,3-dihydro-1-oxo-1H,5H-pyrazolo[1,2-a]pyrazoles 5 by stereoselective cycloadditions of dimethyl acetylene-dicarboxylate (3) to (1Z)-rel-(4R,5R)-1-arylmethylidene-4-benzoylamino-5-phenyl-3-pyrazolidinone-1-azomethine imines 4 [10]. In this paper we report on the transformation of pyrazolo[1,2-a]pyrazoles 5 into rel-(2R,3R)-R-benzoyl-3-phenyl-3-[5-aryl-3,4-bis(methoxycarbonyl)pyrazolyl-1]-alanine methyl esters 6 by oxidative ring opening reaction.

position to give intermediates 7, or by the addition to C=C double bond giving adducts 8. However, regardless of the site of bromination, nucleophilic attack of methanol and elimination of hydrogen bromide lead to the same type of products 6 (Scheme 1).

The structures of 3-phenyl-3-pyrazolylalanine derivatives 6 were confirmed by nmr characterisation and elemental analyses. From a stereochemical point of view, one center of chirality is lost during this transformation. However, since the reaction does not take place at the other

Table 1
Experimental and Analytical Data

Compound Analyses	yield (%)	mp °C	Molecular Formula
6a	87	91-92	$C_{30}H_{27}N_3O_7$
		(from cyclohexane)	Calcd.: C, 66.54; H, 5.03; N, 7.76 Found: C, 66.43; H, 4.87; N, 8.04
6b 6c	94	90-92	$C_{31}H_{29}N_3O_7$
		(from cyclohexane)	Calcd.: C, 67.02; H, 5.26; N, 7.56 Found: C, 67.09; H, 5.10; N, 7.95
	93	88-90	Found: C, 67.09, H, 5.10, N, 7.95 $C_{31}H_{20}N_3O_8$
	93	(from cyclohexane)	Calcd.: C, 65.14; H, 5.11; N, 7.35 Found: C, 65.08; H, 5.10; N, 7.61
6d	83	93-95	C ₃₀ H ₂₆ N ₄ O ₉
		(from cyclohexane)	Calcd.: C, 61.43; H, 4.47; N, 9.55 Found: C, 61.59; H, 4.08; N, 9.79
6e	86	99-101	$C_{30}H_{26}N_4O_9$
		(from cyclohexane)	Calcd.: C, 61.43; H, 4.47; N, 9.55
		24.22	Found: C, 61.58; H, 4.38; N, 9.69
6f	89	81-82 (from cyclohexane)	C ₃₀ H ₂₅ N ₃ O ₇ Cl ₂ Calcd.: C, 59.03; H, 4.13; N, 6.88 Found: C, 58.90; H, 4.05; N, 7.11

two chiral centers at the positions 2 and 3, the reaction provides access to rel-(2R,3R)-3-phenyl-3-[5-aryl-3,4-

bis(methoxycarbonyl)pyrazolyl-1]alanine esters 6 with known relative configuration.

Table 2 1H NMR Data

Compound	¹ H NMR (δ - TMS)
6a	3.68 (3H, s, 4'-COOMe), 3.75 (3H, s, 1-COOMe), 4.00 (3H, s, 3'-COOMe), 5.52 (1H, t, 2-H), 6.00 (1H, d, 3-H), 6.82 (1H,
	d, NH), 7.08-7.70 (15H, m, 3Ph), $J_{H2H3} = J_{CHNH} = 7.0 \text{ Hz}$
6b	2.36 (3H, s, CH ₃ -C ₆ H ₄), 3.65 (3H, s, 4'-COOMe), 3.72 (3H, s, 1-COOMe), 3.97 (3H, s, 3'-COOMe), 5.47 (1H, t, 2-H),
	5.98 (1H, d, 3-H), 6.80 (1H, d, NH), 7.08-7.70 (14H, m, 10H-Ph, C_6H_4), $J_{H2H3} = J_{CHNH} = 7.0 \text{ Hz}$
6c	3.67 (3H, s, 4'-COOMe), 3.72 (3H, s, 1-COOMe), 3.83 (3H, s, CH ₃ O-C ₆ H ₄), 3.98 (3H, s, 3'-COOMe), 5.49 (1H, t, 2-H),
	6.00 (1H, d, 3-H), 6.86 (2H, d, 2H-C6H4), 6.95 (1H, d, NH), 7.11 (2H, d, 2H-C6H4), 7.30-7.54 (10H, m, 2Ph), JH2H3 =
	$J_{CHNH} = 6.8 \text{ Hz}, J_{o,m-C6H4} = 8.4 \text{ Hz}$
6d	3.70 (3H, s, 4'-COOMe), 3.76 (3H, s, 1-COOMe), 4.00 (3H, s, 3'-COOMe), 5.50 (1H, t, 2-H), 6.00 (1H, d, 3-H), 6.78 (1H, d, NH),
	7.30-7.69 (12H, m, 10H-Ph, 2H- C_6H_4), 8.03-8.09 (1H, m, 1H- C_6H_4),), 8.24-8.47 (1H, m, 1H- C_6H_4), $J_{H2H3} = J_{CHNH} = 7.0 \text{ Hz}$
6e	3.68 (3H, s, 4'-COOMe), 3.75 (3H, s, 1-COOMe), 4.00 (3H, s, 3'-COOMe), 5.48 (1H, t, 2-H), 6.00 (1H, d, 3-H), 6.75 (1H, d, NH),
	7.30 (5H, s, Ph), 7.40 (2H, d, 2H-C ₆ H ₄), 7.50 (5H, s, Ph), 8.27 (2H, d, 2H-C ₆ H ₄), $J_{H2H3} = J_{CHNH} = 7.2 \text{ Hz}, J_{o.m-C6H4} = 9.0 \text{ Hz}$
6f	3.63 (3H, s, 4'-COOMe), 3.79 (3H, s, 1-COOMe), 4.01 (3H, s, 3'-COOMe), 5.46 (1H, dd, 2-H), 5.80 (1H, d, 3-H), 7.10 (1H, d, NH),
	7.28-7.82 (13H, m, 10H-Ph, 3H- C_6H_3), $J_{H2H3} = 4.8$ Hz, $J_{CHNH} = 7.0$ Hz

EXPERIMENTAL

Melting points were taken on a Kofler micro hot stage. The ¹H nmr spectra were obtained on a Varian E-360 (60 MHz) spectrometer with deuteriochloroform as solvent and tetramethylsilane as internal standard. The microanalyses for C, H, and N were obtained on a Perkin-Elmer CHN Analyser 2400. rel-(2R,3R,5S)-5-Aryl-6,7-bis(methoxycarbonyl)-2,3-dihydro-1-oxo-1H,5H-pyrazolo[1,2-a]pyrazoles 5a-f were prepared according to the procedures described in the literature [10]. Crude rel-(2R,3R)-N-benzoyl-3-[5-aryl-3,4-bis(methoxycarbonyl)pyrazolyl-1]-3-phenylalanine methyl esters 6a-f were purified by crystallisation from cyclohexane.

rel-(2*R*,3*R*)-*N*-Benzoyl-3-[5-aryl-3,4-bis(methoxycarbonyl)pyrazolyl-1]-3-phenylalanine Methyl Esters **6a-f**. General Procedure.

A mixture of rel-(2R,3R,5S)-5-aryl-6,7-bis(methoxycarbonyl)-2,3-dihydro-1-oxo-1H,5H-pyrazolo[1,2-a]pyrazole (5a-f, 0.001 mole), methanol (4 ml), and bromine (0.06 ml, 0.00117 mole) was heated at reflux temperature for 30 minutes and volatile components evaporated in vacuo. Methanol (2 ml), triethylamine (0.28 ml, 0.002 mole), and water (10 ml) were added to the residue, and the precipitate collected by filtration to give 6a-f, respectively. Experimental and analytical data for rel-(2R,3R)-N-benzoyl-3-[5-aryl-3,4-bis(methoxycarbonyl)-pyrazolyl-1]-3-phenylalanine methyl esters 6a-f are given in Tables 1 and 2.

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REFERENCES AND NOTES

- [1] For a review see: P. Kolar, A. Petrič, and M. Tišler, J. Heterocyclic Chem., 34, 1067 (1997).
- [2] N. Sugimoto, H. Watanabe, and A. Ide, *Tetrahedron*, 11, 231 (1960).
 - [3] I. L. Finar and K. Utting, J. Chem. Soc., 5272 (1960).
- [4] H. Reimlinger and J. F. M. Oth, Chem. Ber., 97, 331 (1964).
- [5] L. D. Arnold, R. G. May, and J. C. Vedras, J. Am. Chem. Soc., 110, 2237 (1988).
- [6] J. E. Baldwin, A. C. Spivey, and C. J. Schofield, Tetrahedron Asymmetry, 1, 881 (1990).
 - [7] M. Perez and R. Pleixats, *Tetrahedron*, **51**, 8355 (1995).
- [8] S. H. Rosenberg, K. P. Spina, K. W. Woods, J. Polakowski, D. L. Martin, Z. Yao, H. H. Stein, J. Cohen, J. L. Barlow, D. A. Egan, K. A. Tricarico, W. R. Baker, and H. D. Kleinert, J. Med. Chem., 36, 449 (1993).
- [9] S. H. Rosenberg, K. P. Spina, S. L. Condon, J. Polakowski, Z. Yao, P. Kovar, H. H. Stein, J. Cohen, J. L. Barlow, V. Klinghofer, D. A. Egan, K. A. Tricarico, T. J. Perun, W. R. Baker, and H. D. Kleinert, J. Med. Chem., 36, 460 (1993).
- [10] J. Svete, A. Prešeren, B. Stanovnik, L. Golič, S. Golič-Grdadolnik, J. Heterocyclic Chem., 34, 1323 (1997).